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Synthetic Studies on 6-Phenyl-Quino Carbazole

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ABSTRACT: Much interest centres around carbazoles because of the airn high level of pharmacological properties.Carbazoles derivaties in general,pyridocarbazoles in particular,(Elliticine(1) and Olivacine(2) exhibit anti – cancer properties.



In the present case, the reaction 0f 1-oxo -1,2,3,4 –tetrahydro carbazolesnwith o-amino benzophenon in the presence of acids as catalyst is taken as objective. Then the resulting 7,8 –dihydro derivative of carbazoles can be used as synthons to derive quino(2,3-b) carbazoles using either Pd|C in diphenyl ether or DDQ in benzene. The is outline below



All the compounds have been characterized by IR ,NMR and elemental analysis. Keywords: Carbazoles derivatives,1-oxo -1,2,3,4 tetrahydrocarbazoles, 7,8-dihydro- 6-phenyl – quino (2,3-a)carbazole using Pd/C.

I. INTRODUCTION

Nitrogen hetero cyclic compounds play vitol role in the metabolism of all living cells, which are widely distributed in nature and are essential to life.

Carbazole derivatives represent a new interesting variant in the large number of indole alkolids, which have yielded several important drugs. Several reports have appeared on the synthesis of carbazoles derivatives, in the



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connection with the search for newer physiologically active compounds. Chowdhury, et al¹, have studied the toxicity and structure activity relationships of fourteen carbazoles derivatives. Carbazomycin A and Carbazomycin B have been found to be useful as antibacterial and anti – fungal agents.²⁻⁴ The ochrosia and Aspidosperma 6H- pyrido(4,3-b) carbazole alkalodis, namely, ellipticine is used linically as a drug to trear advanced breast cancer, myeloblastic leukemia and some solid tumors.⁵⁻⁸ Recently 7H –pyrido(4,3-c)ncarbazoles were reported to elicit anti HIVactivity, Antidepressant, Anticonvulsant and diretic, Trypanocidal, Anticancer, etc.,

II. RESULT AND DISCUSSION

The potential precursors,1-oxo-1,2,3,4 – tetrahydrocarbazoles(2) prepared in our laboratory have openednew avenues viz.,mukonine isomers⁴², functionalized for the synthesis of highly carbazoles derivatives isomers58 oxazolocarbazoles⁵⁷,girinimbine 2,2-dimethyl 2HPyrocarbazoles⁵⁹, indolocoumarins⁶⁰, carbozolyloxypropanolaminess^{55,3-acetyl} -2-hydroxy -1-N,Ndiacetylaminocarbazoles⁶¹, benzocarbazoles⁴³, and pyrazinocarbazoles⁴⁹, in excellent yields.

The 1-oxo-1,2,3,4 –tetra hydrocarbazoles(38) thus obtained were utilized as synthons to derive 6-Phenyl –quino(2,3 - a) carbazoles(40).



6- Methyl -1-oxo-1,2,3,4 –tetrahydocarbazole(38a) on treatment with o-aminobenzophenone in glacialacetic acid yielded an orange coloured solid. This on purification using column chromatography over silica gel yielded yellow colour solid melted at 156° C. Its IR spectrum showed strong stretching at 3275cm⁻¹ due to the presence of N-H group. A bane between 2980 and 2895cm⁻¹ was due to C –H stretching's. A group C=N stretching's appeared at 1640 cm⁻¹ Its ¹H- NMR spectrum in CDCl₃ exhibited in various singlet and multiplet band.

The elemental analysis was also good agreement with the proposed molecular formula $C_{26}H_{20}N_2$. All these spectral and analytical details attest the compound as 7,8 – dihydro -10-methyl -6-phenyl-quino(2,3-a) carbazole(39a).



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The above reaction was generalized for other carbazole derivatives38 (b-e) in order to obtain the respective products 39 (b-e)



The spectral and analytical data infer the obtained product as 7,8 –dihydro-6- phenyl –quino(2,3-a) carbaxole(39). Therfore We planned to get 6-phenyl qoino(2,3-a) carbazole(40)from 7,8-dihydro- 6-phenyl – quino (2,3-a)ncarbazole using Pd/C.



7,8 - dihydro- 10- methyl -6-phenyl-quino(2,3-a) carbazole(39a) refluxed with Pd/C

In diphenyl ether for 5 h to yield a dark brown coloured product. This on purification using column chromatography over silica gel yielded a yellow coloured solid melted at 162° C .Its IR spectrum showed a strong stretching at 3422cm⁻¹ showed the presence of C –H stretching. A stretching appeared at 1667cm⁻¹ was due to C=N group. The product is well supported by its mass spectrum. The molecular ion peak appeared at m/e 358. The elemental analysis agreed well with the assigned structure 10-methyl-6-phenyl-quino(2,3-a)carbazole(40a). The sample has been submitted for its ¹H- NMR specral data.



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III. EXPERIMENTAL

General:

The solvent and reagents used for the preparations were grade and were purified by standard methods, petroleum ether used was of boilind boint range $60-80^{\circ}$ C. Anhydrous sodium sulphate as used to dry the solution of organic extracts.

Thin layer chromatography was performed using glass plates coate with silica gel –G containing 13% calcium sulphate as binder. Ethyl acetate and petroleum ether were used as developing solvents. A chamber containing iodine vapour was used to locate the spots.

Preparation of 2-hydroxymethylencyclohexanone:

Cyclohexnaone(51.7mL, 0.50mol) was added in protion over a period of 5 min.to a well cooled vigorously stirred mixture of sodium methoxide(12.25g of sodium in 125mL of methnol),dry ether(100mL) and ethyl formate (45mL,0.50mol).The mixture was stirred in the ice bath for another 1 h and then allowed to stand at room temperature nfor 24 h. At the end of the period, ice aand water were added tom the yellow solid mass and acidified with cold concentrated hydrocholoric acid. The oil thus separated was extracted with ether,washed with coldwater and brine solution and dried over anhydrous sodium sulphate . the residual oil after the removal of solvent was distilled under reduced pressure togive2hydroxymethylencuclohexanone.

b.p: 83-85⁰C/ 8mm

Yiels: 35g(60%)

Preparation of cyclohexanone-1,-dione -1-arylhydrazones.(37): General procedure:

A solution of appropriate aniline derativatre (0.1mol) in aqueous hydrochloric acid (27 ml of HCL in 53 mL solution) was diazotized with cold saturated aqueous solution of sodium nitite(8.7 g in 20mL water) while the temperature was kept at 0° C to 5° C. The solution waskept aside for 10nminutes. It was then added portion wise to an ice cold mixture containing 2-hydroxymethylencuclohexanone(38), 12.6 g, 0.1 mol), sodium acetate trihydrate(25g), methanol(150mL) and water (75ml) over a period of 0.5hn with nstirring. The contents were allowe top stand for further 0.5 h. and the resulting solid mass was filtered, washed with water, dried and crystallized from methanol. Finally we got 37a, 37b, 37c, 37d & 37e.

CYCLISATION OF THE HYDRAZONE(37):

General procedure:

A solution of the appropriate hydrazone(37,0.001 mol) in a mixture of acetic acid(20mL) and hydrochloric (5 mL) was refluxed on oil bath preheated to $125-130^{\circ}$ C for 0.5 h. The contents were then cooled and poured into coldwater with stirring. The separated brown solid was purified by column chromatography over silica gel using petroleum ether: ethyl acetate(95:5).



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7-methyl – **1-0xo-1,2,3,4-tetrahydrocarbazoles(38a):**Cyclohexane-1,2-dione-1-p-tolylhydrazone(37b) :2.16 g,Glacial acetic acid and hydrochloric acid:20 mL:5mLM.p. :159

160[°]C, Yield: 1.25g, (63%), **I**R(KBR)cm¹: 3250, 1620, 1550, 1520, 1460, 1325, 1350, 1240, 1120, 1060, 880.

8-methyl – 1-0xo-1,2,3,4-tetrahydrocarbazoles(38c):Cyclohexane-1,2-dione-1-p-tolylhydrazone(37c) :2.16 hvdrochloric acid:20 g.Glacial acetic acid and mL:5m.M.p.:163 ^{163⁰}C, Yield: 1.85g, (57%), IR(KBR)cm¹:3200, 1630, 1600, 1530, 1460, 1370, 1320, 1240, 1220, 1180, 1160, 1120, 1020, 840. 1-oxo-1,2,3,4-tetrahydrocarbazole(38c)Cyclohexane-1,2-dione-1-phenylhydrazone(37c):2 gGlacial acetic acid and hydrochloric acid:20 mL:5mL,M.p:166 168[°]C,Yield:1.2 g(65%),IR(KBR)cm⁻¹ -:3250, 1620, 1550, 1520, 1450, 1320, 1240, 1220, 1170, 1150, 1120, 880.

Reaction of 1-oxo-1,2,3,4-tetrahydrocarbazoles(38) with o-aminobenzophenone: General procedure:

A solution of an 1-oxo-1,2,3,4-tetrahydrocarbazoles(38,0.005 mL) was refluxed for 8 hrs in the presence of a drop of sulphuric acid, The reaction was monitored by TLC.After the completion of the reaction, it was poured into crushed ice and extracted with chloroform and dried over anhydrous sodium sulphate. The crude product thus obtained on removal of the solvent, was purified by bcolumn chromatography over silica gel using per.ether: ethyl acetate(98:5) to yieldthe corresponding product.

7,8- Dihydro – 6-phenyl –quino(2,3-a) carbazole(39e):1-oxo 1,2,3,4

 $tetrahydrocarbazole(38e): 0.925g(0.005mol), oaminobenzophenone: 0.985g(0.005mL)M.p:>300^{0}C, Yield: 0.573g(62\%), IR(KBR)V_{max}Cm^{1}: 3413, 2922, 2360, 1639, 1548, {}^{1}HNMR(CDcl_{3)}^{\delta}12.30(b s, 1H, N_{13}-H), 8.57-6.96(m, 13H, C_{2}^{-}, C_{3}^{-}, C_{4}^{-}, C_{5}^{-}, C_{9}^{-}, C_{10}^{-}, C_{11}^{-}, C_{12}^{-}, C_{2}^{-}, C_{3}^{-}, C_{4}^{-}, C_{5}^{-}, C_{6}^{-}-H), 3.08-2.93(m, 4H, C_{7}-2H, C_{8-2H}),$

Reaction of 7,8-dihydro -6-phenyl –quino (2,3-b) carbazole(39,0.001mol) Pd/C(0.050)in 6 ml of diphenyl ether was refluxed for about 5 h. The reaction was monitored by TLC. It was then filtered by Whaatmann-41 filter paper to remove Pd/c. After the removal of solvent, the mixture was poured into crushed ice, extracted using ethyl acetate dried over anhydrous sodium sulphate. The crude product, Thus obtained was purified by column chromatography over silica gel pet.ether: ethyl acetate(98:2) to yield the corresponding 6 - phenyl - quino(2,3-a) carbazole(40).

| Analysis:C ₂₆ H ₁₈ N ₂ | C | Н | Ń |
|---|-------|------|------|
| Calculated | 87.12 | 5.06 | 7.82 |
| Found | 87.23 | 5.01 | 7.76 |

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